## THE RELATIONS BETWEEN TORSIONAL AND VALENCY ANGLES OF CYCLOPENTANE

## H.J. Geise, C. Altona and C. Romers

Laboratory of Organic Chemistry, The University of Leiden, Netherlands

## (Received 1 February 1967)

In 1960 Hendrickson (1) computed the energies of various cyclopentane conformers as a function of geometric parameters, i.a. torsional and valency angles. The aim of the present note is to correlate the results of his calculations with the conclusions of Pitzer et al. (2,3) concerning the pseudo-rotation of cyclopentane. Pitzer et al. described the cyclopentane ring as having a certain degree of puckering which rotates around the ring and which is associated with a phase angle P. Their calculations are based upon the assumption that the displacement  $z_j$  of the j th atom from the plane of the unpuckered ring can be written as

$$\mathbf{z}_{j} = \sqrt{\frac{2}{5}} q \cos \left\{ 2(P + \frac{2\pi j}{5}) \right\}$$
(1)

where q is the amplitude of the puckering. Their calculations are restricted to such values of P for which the molecule has  $C_g$  or  $C_2$  symmetry and the authors argued that the energy depends largely on the value of q (the degree of puckering) and only slightly on P; i.e. the potential barrier in the pseudo-rotation circuit is very small.

It was thought worthwhile to have at our disposal relations which afford all torsional angles  $\varphi_j$  and valency angles  $\vartheta_j$  as a function of P given a certain degree of puckering (consult Fig. 1a for the numbering of the various angles). We plotted Hendrickson's corrected values (4) for  $\varphi_j$  against those for  $\vartheta_{j+1}^2$  (i = 1,2,3,4). (Fig. 1b) and found that in the region where the total energy is constant, the torsional angles are related by

$$P_{j} = \underline{a} \sin \left( P + \frac{4\pi j}{5} \right); \quad j = 1, 2....5$$
 (2)

The constant <u>a</u> is the amplitude of the vibrational angular motion depicted in Fig. 1b. A similar plot of  $V_j$  versus  $V_{j+1}$  reveals that  $V_j$  can be expressed as  $V_j = 7 + b \sin \left\{ 2 \left( P + \frac{4\pi j}{5} \right) + 5 \right\}; \quad j = 1, 2...5$  (3)

in the same region, indicating that the valency angles ''vibrate'' between the values (T - b and T + b). The phase lag can be evaluated from a plot of  $\mathcal{V}_j$  versus  $\mathcal{V}_j$  (Fig. 1c). Noting that the cyclopentane ring has  $C_g$  symmetry for P = 0 when  $\mathcal{F}_5 = 0$  and  $\mathcal{V}_1 = \mathcal{V}_5$ ,

it follows that  $\delta = 7/10 \pi$  or  $-3/10 \pi$ . The plots depicted in Fig. 1 are Lissajous curves and clearly demonstrate that the angular motions have a vibrational character in the region of minimum energy. The parameters a,  $\tau$  and b (based upon Hendrickson's calculations) have respective values of 44°, 104° and 2.5°.

The following conclusions can be drawn:

1°. Up to  $\psi_1 = 40.0^\circ$  the table published by Hendrickson (4) represents a large part of the pseudo-rotation itinerary of one cyclopentane model, whereas the other entries correspond to a number of cyclopentane models differing in degree of puckering at a certain point of the rotational pathway.

- 2°. Throughout the pseudo-rotation circuit (by varying P between 0 and 27) 10C<sub>g</sub> or envelope forms and 10 C<sub>2</sub> or half-chair forms are met. Torsional angle  $\varphi_j$  about bond  $C_j - C_{j+1}$  as well as valency angle  $\vartheta_{j+3}^j$  attain their maximum values when a twofold rotation axis passes through  $C_{j+3}$  and the midpoint of  $C_j - C_{j+1}$ . Torsional angle  $\vartheta_j$  is zero and valency angle  $\vartheta_{j+3}^j$  attains its minimum value when a mirrorplane goes through  $C_{j+3}$  and is perpendicular to the line  $C_j - C_{j+1}$  in its midpoint.
- 3°. Experimental knowledge of two torsional angles is sufficient for calculation of all other angular magnitudes.
- 4°. It is not possible to choose ,  $\mathcal{T}$  and b independently. A certain degree of puckering i.e. a given value of a implies a corresponding pair  $\mathcal{T}$ , b. Substituting eqs. (2) and

(3) in Pitzer and Donath's (3) geometrical relations:  $\cos \varphi_2 = \left[\cos \vartheta_2 \left(\cos \vartheta_3 - 1\right) - \sin 1/2 \vartheta_3 \left(1 - 2\sin 1/2 \vartheta_3\right)\right] / \sin \vartheta_2 \sin \vartheta_3,$   $\cos \varphi_1 = \left[\cos \vartheta_2 \cos \vartheta_1 + \sin 1/2 \vartheta_3\right] / \sin \vartheta_2. \sin \vartheta_1,$ a set of equations is obtained from which  $\tau$  and b can be solved (Fig. 2).

In following papers of this series (5,6) we will show that the spread in  $a, \mathcal{I}$  and b, even for substituted cyclopentane rings occurring in steroids, is small. In steroids we can make use of a ''standard'' ring D with a precision of about 1°. For the weighted average of the calculated <u>a</u> value of the five membered ring in 11 steroids (6) or steroidlike compounds we find  $46.7^{\circ} \stackrel{+}{-} 0.7^{\circ}$ . The standard torsional angles of ring in the two symmetrical forms are compared below with theoretical values for a free cyclopentane ring:

A . A		C <sub>s</sub> - form		
theor <sup>1,4</sup>	theor <sup>3</sup>	exp <sup>6</sup>	theor <sup>1,4</sup>	theor <sup>3</sup>
7 + 44.0	+ 48.2	+ 44.4	+ 41.8	+ 45.8
3 – 35 <b>.5</b>	- 38.8	- 44.4	- 41.8	- 45.8
4 + 13.7	+ 15.1	+ 27.4	+ 26.0	+ 28.7
4 + 13.7	+ 15.1	0	0	0
3 - 35.5	- 38.8	- 27.4	- 26.0	- 28.7
	$\frac{\text{theor}^{1,4}}{7 + 44.0}$ $3 - 35.5$ $4 + 13.7$ $4 + 13.7$ $8 - 35.5$	$\frac{\text{theor}^{1,4}}{7} + 44.0 + 48.2$ $3 - 35.5 - 38.8$ $4 + 13.7 + 15.1$ $4 + 13.7 + 15.1$ $8 - 35.5 - 38.8$	theor <sup>1,4</sup> theor <sup>3</sup> $\exp^6$ 7+ 44.0+ 48.2+ 44.43- 35.5- 38.8- 44.44+ 13.7+ 15.1+ 27.44+ 13.7+ 15.108- 35.5- 38.8- 27.4	theor $1.4$ theor $exp^6$ theor $1.4$ 7+ 44.0+ 48.2+ 44.4+ 41.83- 35.5- 38.8- 44.4- 41.84+ 13.7+ 15.1+ 27.4+ 26.04+ 13.7+ 15.1008- 35.5- 38.8- 27.4- 26.0

The authors thank Prof. Havinga for his interest during this investigation.

## References.

1) J.B. Hendrickson, <u>J.Am.Chem.Soc</u>. <u>83</u>, 4537 (1961)

- 2) J.E. Kilpatrick, K.S. Pitzer and R. Spitzer, J.Am.Chem.Soc. 69, 2483 (1947)
- 3) K.S. Pitzer and W.E. Donath, <u>J.Am.Chem.Soc</u>. <u>81</u>, 3213 (1959)
- 4) J.B. Hendrickson, <u>J.Am.Chem.Soc</u>. <u>85</u>, 4059 (1963)
- 5) H.J. Geise, C. Altona and C. Romers, Tetrahedron, 23, 439 (1967)
- 6) C. Altona, H.J. Geise and C. Romers, Tetrahedron, to be published.



Fig. 1. a) Numbering of torsional and valency angles in cyclopentane; b) Plots of  $\mathscr{P}_{\mathbf{j}}$  vs  $\mathcal{P}_{i}$ ; c) Plots of  $\mathcal{V}_{j}$  vs  $\mathcal{V}_{i}$ ; d) Plot of  $\mathcal{P}_{j}$  vs  $\mathcal{V}_{j}$ .

